

**Discussion.** The final atomic parameters for the title compound are listed in Table 1.\* A perspective view of the molecule and the atom-numbering scheme are given in Fig. 1. Fig. 2 shows the structural formula, labels of C—C bonds, and dimensions averaged according to idealized  $D_{2h}$  molecular symmetry.

The sets of atoms (I) [C(4), C(5), C(7) and C(8)], (II) [C(16), C(17), C(19) and C(20)], (III) [C(1), C(2), and C(21) to C(24)] and (IV) [C(9) to C(14)] all conform to strict planarity, the maximum deviations from the corresponding best planes being 0.002, 0.003, 0.009 and 0.009 Å, respectively. The molecule, however, appears to deviate very slightly but perceptibly from rigorous  $D_{2h}$  symmetry. The dihedral angles ( $\sigma \sim 0.3^\circ$ ) between pairs of planes are: (I) and (II),  $0.7^\circ$ ; (I) and (III),  $87.7^\circ$ ; (II) and (III),  $87.8^\circ$ ; (I) and (IV),  $87.9^\circ$ ; (II) and (IV),  $88.0^\circ$ ; (III) and (IV),  $1.3^\circ$ .

The considerable molecular strain is reflected by the fact that each *para*-fused benzenoid ring is boat-shaped, with its bow and stern C atoms displaced inward towards the other ring by 0.15 (1) Å. The transannular non-bonded distances  $p$  and  $q$  (see Fig. 2) agree closely with the corresponding values of 2.78 and 3.09 Å found in [2.2]paracyclophane (Hope, Bernstein & Trueblood, 1972). The inter-ring bridging bond  $e$  is normal (see data compiled by Keehn, 1983) and in

\* Lists of bond lengths and bond angles, anisotropic thermal parameters and H-atom coordinates have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 42735 (11 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

excellent accord with the measured distance of 1.488 (2) Å in 1,2:3,4:11,12:13,14-tetrabenz[4.4]-metacyclophane (Irngartinger, 1976). An unusual feature in the present structure is the significant lengthening of bond  $d$  in relation to other C—C bonds in the *ortho*-fused ring. A precedent of this type of distortion in highly strained hydrocarbons exists in 5,6,11,12-tetrahydrodibenzo[*a,e*]cyclooctene, in which the inner  $C_{sp^2}-C_{sp^2}$  bond has a length of 1.435 (2) Å at 290 K (Destro, Pilati & Simonetta, 1977).

#### References

- CHAN, C. W. & WONG, H. N. C. (1985). *J. Am. Chem. Soc.* **107**, 4790–4791.  
 DESTRO, R., PILATI, T. & SIMONETTA, M. (1977). *Acta Cryst.* **B33**, 447–456.  
 DETITTA, G. T., EDMONDS, J. W., LANGS, D. A. & HAUPTMAN, H. (1975). *Acta Cryst.* **A31**, 472–479.  
 HOPE, H., BERNSTEIN, J. & TRUEBLOOD, K. N. (1972). *Acta Cryst.* **B28**, 1733–1743.  
*International Tables for X-ray Crystallography* (1974). Vol. IV, pp. 55, 99, 149. Birmingham: Kynoch Press. (Present distributor D. Reidel, Dordrecht.)  
 IRNGARTINGER, H. (1976). *Acta Cryst.* **B32**, 696–702.  
 JACOBSON, N. & BOEKELHEIDE, V. (1978). *Angew. Chem. Int. Ed. Engl.* **17**, 46–47.  
 KEEHN, P. M. (1983). *Cyclophanes*, Vol. I, edited by P. M. KEEHN & S. M. ROSENFELD, pp. 69–238. New York: Academic Press.  
 SCHILLING, J. W. (1970). *Crystallographic Computing*, edited by F. R. AHMED, pp. 201–204. Copenhagen: Munksgaard.  
 SHELDRICK, G. M. (1982). *Computational Crystallography*, edited by D. SAYRE, pp. 506–514. Oxford Univ. Press.  
 SPARKS, R. A. (1976). *Crystallographic Computing Techniques*, edited by F. R. AHMED, pp. 452–467. Copenhagen: Munksgaard.

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### Structure of 4 $\beta$ ,10 $\beta$ -Dimethyleudesma-5,11(13)-diene-7 $\beta$ -lactone (Alantolactone)\*

BY HELMUT W. SCHMALLE

Mineralogisch-Petrographisches Institut der Universität Hamburg, Grindelallee 48, D-2000 Hamburg 13, Federal Republic of Germany

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**Abstract.**  $C_{15}H_{20}O_2$ ,  $M_r = 232.32$ , orthorhombic,  $P2_12_12_1$ ,  $a = 6.258$  (1),  $b = 8.156$  (1),  $c = 25.338$  (1) Å,  $V = 1293.3$  (2) Å<sup>3</sup>,  $Z = 4$ ,  $D_x = 1.193$  Mg m<sup>-3</sup>, Cu  $K\alpha$  radiation,  $\lambda = 1.5418$  Å,  $\mu = 0.53$  mm<sup>-1</sup>,  $F(000) = 504$ ,  $T = 296$  K, final  $R = 0.042$  for 663 observed reflections. The naturally occurring

sensitizing sesquiterpene lactone exhibits a distorted chair-shaped cyclohexane ring which is connected with a cyclohexane ring showing a deformed boat conformation. The 7,8-*cis*-fused lactone ring has a flattened envelope form. C(8) deviates  $-0.12$  (1) Å, the  $\alpha$ -methylene C(13)  $0.11$  (1) Å and the carbonyl oxygen O(2)  $-0.01$  (1) Å from the best plane. Bond distances and angles are normal. The structure consists of discrete molecules.

\* IUPAC name: 3 $\alpha,5,6,7,8,8a,9,9a\alpha$ -octahydro-5 $\beta,8a\beta$ -dimethyl-3-methylenaphtho[2,3-*b*]furan-2(3*H*)-one.

**Introduction.** High allergy-inducing activity of helenin – a mixture of the sesquiterpene lactones alantolactone and isovalantolactone, isolated from *Inula helenium* L. (Family Asteraceae) – has been observed by Hjorth (1970). Stampf, Benezra, Klecak, Geleick, Schulz & Hausen (1982) confirmed this finding in their investigation of helenin and its separated constituents. Stampf, Schlewer, Ducombs, Foussereau & Benezra (1978) reported that alantolactone and isovalantolactone were sensitizers on human volunteers and on guinea pigs, as well. Bleumink, Mitchell, Geissman & Towers (1976) observed that patients suffering from chrysanthemum contact allergy show positive allergic skin reactions to alantolactone. Similar observations have been made on guinea pigs sensitized with chrysanthemum extract (Schulz, Hausen, Wallhöfer & Schmidt-Löffler, 1975). Since cross reactivities were observed between alantolactone – which is used as a standard allergenic test compound – and different sesquiterpene lactone containing plant extracts, X-ray structure determinations of the title compound, isovalantolactone (Schmalle, Schoppe, Jarchow, Hausen & Schulz, 1986), and diepoxy-yomogin (Schmalle, Klupsch, Jarchow, Hausen & Schulz, 1986) have been performed in order to compare their conformations with those of other sesquiterpene lactones causing contact allergy.

**Experimental.** Naturally occurring alantolactone and isovalantolactone from *Inula helenium* L., separated by thin-layer chromatography; transparent needle-shaped crystals grown by slow evaporation from *n*-hexane and ethyl acetate; space-group determination from Weissenberg photographs; crystal dimensions 0.06 × 0.11 × 0.24 mm, Enraf–Nonius CAD-4 diffractometer; cell dimensions measured by least-squares refinement of 19 reflections in the interval 11° <  $\theta$  < 26°; graphite-monochromatized Cu K $\alpha$  radiation; max.  $\sin\theta/\lambda = 0.518 \text{ \AA}^{-1}$ ,  $\theta$ -2 $\theta$  scan, zigzag mode, variable scan rate 0.3° to 20.0° min<sup>-1</sup>,  $-6 \leq h \leq 0$ ,  $-8 \leq k \leq 8$ ,  $0 \leq l \leq 26$ ; 10% loss of intensities in standard reflections  $0\bar{2}6$  and  $0,0,\bar{1}0$  monitored at the beginning and every 2 h (24 times) during data collection; 1654 reflections measured; data reduction with program of CAD-4 structure-determination package; Lorentz-polarization correction, no absorption correction;  $R_{\text{int}} = 0.031$ ; 944 unique reflections, 663 with  $I > 3\sigma(I)$ , 281 unobserved. Structure solved by direct methods with MULTAN80 (Main, Fiske, Hull, Lessinger, Germain, Declercq & Woolfson, 1980) and refined on  $F_{\text{hkl}}$  by full-matrix least-squares technique; all H-atom positions subsequently determined from  $\Delta\rho$  and refined with fixed  $B_{\text{iso}} = 3.16 \text{ \AA}^2$ ; number of reflections in final least-squares cycle  $m = 663$ , parameters refined,  $n = 214$ , unit weight,  $R = 0.042$ ;  $(\Delta/\sigma)_{\text{max}} = 0.65$  (0.24 for non-H atoms) in final refinement cycle; max. and min. heights in final  $\Delta\rho$  0.14 and  $-0.15 \text{ e \AA}^{-3}$ ; computer programs used: MULTAN80 (Main *et al.*,

1980), SHELX76 (Sheldrick, 1976), XANADU (Roberts & Sheldrick, 1975), ORTEP (Johnson, 1971), ORFFE (Busing, Martin, Levy, Brown, Johnson & Thiessen, 1971); atomic scattering factors from SHELX76. The final atomic parameters are given in Table 1.\*

\* Lists of structure factors, anisotropic thermal parameters of the C and O atoms, H-atom coordinates and bond lengths and angles concerning the H atoms have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 42733 (10 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

Table 1. Final atomic coordinates and  $U_{\text{eq}}$  values for alantolactone with e.s.d.'s in parentheses

$U_{\text{eq}}$  values are given in the form  $\frac{1}{3}\sum_i \sum_k U_{ik} a_i^* a_k^*$  ( $a_i, a_k$ ).

|       | x          | y          | z          | $U_{\text{eq}}(\text{\AA}^2)$ |
|-------|------------|------------|------------|-------------------------------|
| O(1)  | 0.8200 (9) | 0.5005 (6) | 0.5131 (2) | 0.054 (3)                     |
| O(2)  | 1.1423 (9) | 0.3956 (6) | 0.4984 (2) | 0.067 (4)                     |
| C(1)  | 0.534 (1)  | 0.8841 (9) | 0.6271 (3) | 0.053 (5)                     |
| C(2)  | 0.608 (2)  | 0.960 (1)  | 0.6787 (3) | 0.065 (6)                     |
| C(3)  | 0.608 (2)  | 0.835 (1)  | 0.7230 (3) | 0.060 (5)                     |
| C(4)  | 0.755 (1)  | 0.6897 (9) | 0.7100 (3) | 0.051 (5)                     |
| C(5)  | 0.717 (1)  | 0.6194 (8) | 0.6556 (2) | 0.039 (4)                     |
| C(6)  | 0.729 (1)  | 0.4605 (9) | 0.6477 (2) | 0.050 (5)                     |
| C(7)  | 0.728 (1)  | 0.3733 (8) | 0.5957 (3) | 0.048 (5)                     |
| C(8)  | 0.643 (1)  | 0.4802 (9) | 0.5499 (3) | 0.049 (5)                     |
| C(9)  | 0.553 (1)  | 0.645 (1)  | 0.5665 (3) | 0.050 (5)                     |
| C(10) | 0.678 (1)  | 0.7363 (7) | 0.6096 (2) | 0.040 (4)                     |
| C(11) | 0.943 (1)  | 0.3241 (8) | 0.5767 (3) | 0.043 (4)                     |
| C(12) | 0.987 (2)  | 0.4059 (9) | 0.5255 (3) | 0.052 (5)                     |
| C(13) | 1.093 (2)  | 0.233 (1)  | 0.6003 (3) | 0.070 (6)                     |
| C(14) | 0.888 (2)  | 0.8019 (9) | 0.5881 (3) | 0.050 (5)                     |
| C(15) | 0.990 (2)  | 0.730 (1)  | 0.7194 (3) | 0.074 (7)                     |

Table 2. Bond lengths (Å) and angles (°) of alantolactone

E.s.d.'s are given in parentheses.

|                 |           |                   |           |
|-----------------|-----------|-------------------|-----------|
| O(1)–C(8)       | 1.457 (8) | C(5)–C(10)        | 1.525 (8) |
| O(1)–C(12)      | 1.336 (9) | C(6)–C(7)         | 1.497 (9) |
| O(2)–C(12)      | 1.193 (8) | C(7)–C(8)         | 1.55 (1)  |
| C(1)–C(2)       | 1.52 (1)  | C(7)–C(11)        | 1.48 (1)  |
| C(1)–C(10)      | 1.569 (9) | C(8)–C(9)         | 1.52 (1)  |
| C(2)–C(3)       | 1.52 (1)  | C(9)–C(10)        | 1.536 (9) |
| C(3)–C(4)       | 1.54 (1)  | C(10)–C(14)       | 1.52 (1)  |
| C(4)–C(5)       | 1.512 (9) | C(11)–C(12)       | 1.485 (9) |
| C(4)–C(15)      | 1.53 (1)  | C(11)–C(13)       | 1.34 (1)  |
| C(5)–C(6)       | 1.314 (9) |                   |           |
| C(2)–C(1)–C(10) | 112.4 (7) | C(7)–C(8)–O(1)    | 106.4 (6) |
| C(1)–C(2)–C(3)  | 111.3 (7) | C(9)–C(8)–O(1)    | 111.0 (6) |
| C(2)–C(3)–C(4)  | 111.1 (6) | C(8)–C(9)–C(10)   | 116.0 (6) |
| C(3)–C(4)–C(5)  | 113.2 (7) | C(1)–C(10)–C(5)   | 110.9 (5) |
| C(3)–C(4)–C(15) | 112.2 (8) | C(1)–C(10)–C(9)   | 106.3 (6) |
| C(5)–C(4)–C(15) | 112.1 (7) | C(5)–C(10)–C(9)   | 108.8 (5) |
| C(4)–C(5)–C(6)  | 120.3 (6) | C(1)–C(10)–C(14)  | 109.1 (6) |
| C(4)–C(5)–C(10) | 119.0 (6) | C(5)–C(10)–C(14)  | 110.8 (6) |
| C(6)–C(5)–C(10) | 120.6 (6) | C(9)–C(10)–C(14)  | 110.9 (6) |
| C(5)–C(6)–C(7)  | 127.1 (6) | C(7)–C(11)–C(12)  | 109.3 (7) |
| C(6)–C(7)–C(8)  | 113.2 (6) | C(7)–C(11)–C(13)  | 129.9 (7) |
| C(6)–C(7)–C(11) | 114.2 (7) | C(12)–C(11)–C(13) | 120.7 (8) |
| C(8)–C(7)–C(11) | 102.8 (6) | C(11)–C(12)–O(1)  | 108.7 (7) |
| C(7)–C(8)–C(9)  | 114.8 (6) | C(11)–C(12)–O(2)  | 128.5 (8) |
| O(1)–C(12)–O(2) | 122.9 (7) | C(8)–O(1)–C(12)   | 112.2 (5) |



and the  $\epsilon$ -amino group of lysine has been demonstrated by Dupuis, Mitchell & Towers (1974). Dupuis, Benezra, Schlewer & Stampf (1980) successfully reacted alantolactone with model proteins and with guinea pig skin protein extract.

Intermolecular distances were calculated up to 3.6 Å with ORFFE (Busing *et al.*, 1971). There are no C—H...O interactions between neighbouring molecules; the structure consists of discrete molecules.

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#### References

- BLEUMINK, E., MITCHELL, J. C., GEISSMAN, T. A. & TOWERS, G. H. N. (1976). *Contact Dermatitis*, 2, 81–88.
- BUSING, W. R., MARTIN, K. O., LEVY, H. A., BROWN, G. M., JOHNSON, C. K. & THIESSEN, W. E. (1971). ORFFE. Oak Ridge National Laboratory, Tennessee.
- COCKER, W. & MCMURRY, T. B. H. (1960). *Tetrahedron*, 8, 181–204.
- DUAX, W. L. & NORTON, D. A. (1975). In *Atlas of Steroid Structure*. New York, Washington, London: Plenum.
- DUPUIS, G., BENEZRA, C., SCHLEWER, G. & STAMPF, J.-L. (1980). *Mol. Immunol.* 17, 1045–1051.
- DUPUIS, G., MITCHELL, J. C. & TOWERS, G. H. N. (1974). *Can. J. Biochem.* 52, 575–581.
- HJORTH, N. (1970). *Contact Dermatitis Newslett.* 8, 180.
- JOHNSON, C. K. (1971). ORTEP. Report ORNL-3794, revised. Oak Ridge National Laboratory, Tennessee.
- KARLE, I. L. (1972). *Acta Cryst.* B28, 2000–2007.
- MAIN, P., FISKE, S. J., HULL, S. E., LESSINGER, L., GERMAIN, G., DECLERCQ, J.-P. & WOOLFSON, M. M. (1980). *MULTAN80. A System of Computer Programs for the Automatic Solution of Crystal Structures from X-ray Diffraction Data*. Univs. of York, England, and Louvain, Belgium.
- MARSHALL, J. A. & COHEN, N. (1964). *J. Org. Chem.* 29, 3727–3729.
- MARSHALL, J. A., COHEN, N. & HOCHSTETLER, A. R. (1966). *J. Am. Chem. Soc.* 88, 3408–3417.
- ROBERTS, P. & SHELDRIK, G. M. (1975). *XANADU*. Program for crystallographic calculations. Univ. of Cambridge, England.
- SCHMALLE, H. W. (1986). *Acta Cryst.* In preparation.
- SCHMALLE, H. W., JARCHOW, O. H. & HAUSEN, B. M. (1980). *Acta Cryst.* B36, 2450–2453.
- SCHMALLE, H. W., KLUPSCH, S., JARCHOW, O. H., HAUSEN, B. M. & SCHULZ, K.-H. (1986). *Acta Cryst.* In preparation.
- SCHMALLE, H. W., SCHOPPE, I., JARCHOW, O. H., HAUSEN, B. M. & SCHULZ, K.-H. (1986). *Acta Cryst.* In preparation.
- SCHULTZ, A. G., GODFREY, J. D., ARNOLD, E. V. & CLARDY, J. (1979). *J. Am. Chem. Soc.* 101, 1276–1277.
- SCHULZ, K. H., HAUSEN, B. M., WALLHÖFER, L. & SCHMIDT-LÖFFLER, P. (1975). *Arch. Derm. Forsch.* 251, 235–244.
- SHELDRIK, G. M. (1976). *SHELX76*. Program for crystal structure determination. Univ. of Cambridge, England.
- STAMPF, J.-L., BENEZRA, C., KLECAR, G., GELEICK, H., SCHULZ, K.-H. & HAUSEN, B. M. (1982). *Contact Dermatitis*, 8, 16–24.
- STAMPF, J.-L., SCHLEWER, G., DUCOMBS, G., FOUSSEREAU, J. & BENEZRA, C. (1978). *Br. J. Dermatol.* 99, 163–169.

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## The Stereochemistry of the Major Epimer Formed on Reaction of Methyl 1-Methyl-6-oxo-2,4-cyclohexadiene-1-carboxylate with Dimethyl Acetylenedicarboxylate: Trimethyl *endo*-5-Methyl-6-oxobicyclo[2.2.2]octa-2,7-diene-2,3,*exo*-5-tricarboxylate

BY JEFFERY F. SAWYER, ANABELA GOMES AND PETER YATES

Lash Miller Chemical Laboratories, University of Toronto, Toronto, Ontario, Canada M5S 1A1

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**Abstract.** C<sub>15</sub>H<sub>16</sub>O<sub>7</sub>, *M<sub>r</sub>* = 308.3, monoclinic, *I*2/a, *a* = 19.534 (5), *b* = 7.213 (5), *c* = 21.371 (6) Å,  $\beta$  = 101.53 (2)°, *U* = 2950 (4) Å<sup>3</sup>, *Z* = 8, *D<sub>x</sub>* = 1.39 g cm<sup>-3</sup>, Mo *K*̄ radiation,  $\lambda$  = 0.71069 Å,  $\mu$ (Mo *K*̄) = 1.04 cm<sup>-1</sup>, *F*(000) = 1296, *T* = 298 K, *R* = 0.058 for 1155 reflections with *I* ≥ 3 $\sigma$ (*I*). X-ray analysis indicates that dimethyl acetylenedicarboxylate reacts preferentially on the methyl face of methyl 1-methyl-6-oxo-2,4-cyclohexadiene-1-carboxylate to give trimethyl *endo*-5-methyl-6-oxobicyclo[2.2.2]octa-2,7-diene-2,3,*exo*-5-tricarboxylate. In the bicyclo[2.2.2]octadiene system, opposite bonds are very

slightly twisted away from being eclipsed, with torsion angles about the C(2)=C(3) and C(7)=C(8) double bonds and bond C(5)—C(6) of -2.3 (5), 1.2 (6) and -1.7 (3)°, respectively. The two methoxycarbonyl groups are canted by 12.4 (7) and 79.7 (7)° with respect to the plane of the C(2)=C(3) double bond.

**Introduction.** A key step in the syntheses of cedranoid sesquiterpenes that are being pursued in these laboratories is the photosensitized isomerization of bicyclo[2.2.2]octenones of type (4) to tricyclo[3.2.1.0<sup>2,8</sup>]octanones (5); compounds (4) were prepared